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- Although the melting point and nmr spectrum of this compound do not agree with the reported values,¹² the ir and nmr spectra are identical with the reference spectra provided by Professor Tetsuji (26) Kametani, Tohoku University.

C-Glycosyl Nucleosides. V. A Novel One-Step Asymmetric Synthesis of C-Nucleoside Analogs¹

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Reaction of lithiated heterocycles such as pyridine, benzothiazole, imidazole, benzimidazole, and sydnone with sugar lactones, 2,3:5,6-di-O-isopropylidene-L-gulono-1,4-lactone (3) or 2,3-O-isopropylidene-D-ribono-1,4lactone (7), afforded a variety of 1-(2-substituted heterocyclic)-2,3:5,6-di-O-isopropylidene- β -L-gulofuranose (4) or 1-(2-substituted heterocyclic)-2,3-O-isopropylidene-β-D-ribofuranose (8). Attempted dehydroxygenation of the anomeric hydroxyl group failed. These C-nucleoside analogs were reduced with sodium borohydride to gulitols and ribitols. The configuration of gulitols, which had a π -electron ring system, was determined with CD and ORD spectra to confirm their absolute configuration. It was concluded that a similar Cotton effect is observed in furanose-type and gulitol-type nucleosides.

Synthetic studies on the nucleoside antibiotic, pyrazomycin, have been reported by Tronchet and Perret.² On the other hand, Townsend and his collaborators synthesized its analogous N-nucleoside³ and pyrazolopyrimidine nucleosides.⁴ Several synthetic routes directed to C-nucleosides have also been reported by Fox and Ohrui.⁵ In the previous paper,⁶ we reported the reaction of ethynyl compounds with lactones, and the resulting compound had been expected as an intermediate for the preparation of the carbon-linked nucleoside. In another paper,⁷ we reported the ethynylation of glucosyl bromide with ethynylmagnesium bromide, although we could not obtain the desired carbon-linked nucleoside. The attempted 1,3-dipolar cycloaddition reaction of 1-ethynylphenyl-2,3-O-isopropylidene- α -p-ribofuranose (1) and N-benzylsydnone failed.

The present paper concerns itself with a direct reaction of some lithiated heterocycles with sugar lactones to yield a carbon-linked nucleoside. The reaction of 2,3:5,6-di-Oisopropylidene-L-gulono-1,4-lactone (3) or 2,3-O-isopropylidene-p-ribono-1,4-lactone (7) with various lithiated heterocycles gave gulofuranosyl derivatives (4a-g) or ribofuranosyl derivatives (8b,c).

By application of the reported method 6 of ethynylation with lactones to the reaction of heterocycles with sugar lactones, it has been possible to obtain heterocyclic sugar lactols. Treatment of 3 with *n*-butyllithium and α -bromopyridine, benzothiazole, or 1-benzylbenzimidazole gave 2,3:5,6-di-O-isopropylidenegulonolactols 1-substituted (4a-c) in a good yield (74, 56, and 40%, respectively, Chart I). The ir spectra of these compounds showed hydroxyl bands in the 3200-3380-cm⁻¹ region, and no lactonic band at around 1780 cm⁻¹. Gulonolactols (4a-c) were acetylated with acetic anhydride in pyridine to yield



their acetyl derivatives (5a-c). This result was similar to those of ethynyl derivatives.⁶ In the case of 1-benzylbenzimidazole, the lithiation does occur at the 2 position similar to that of benzothiazole,⁸ and this fact was confirmed from the nmr spectra of 4b. Micetich⁹ reported that lith-

Synthesis of C-Nucleoside Analogs



200 250 300 nm Figure 1. CD (----) and ORD (----) curves in methanol at 28°.



iation of isothiazoles and thiadiazoles gave the 5-lithio compounds. The lithiation of 1-benzylbenzimidazole by n-butyllithium at the 2 position is supported by the formation of 1,1'-dimethyl-2,2'-bibenzimidazole from the lithiation of 1-methylbenzimidazole.¹⁰ On the other hand, treatment of 1-benzylimidazole under the same condition as above afforded 30% of 2-substituted compound 4e and 12% of 5-substituted compound 4f. The nmr spectrum of 2-substituted benzylimidazole (4e) showed a pair of doublets at δ 6.77 and 6.98 ppm (J = 5 Hz) corresponding to H-4 and H-5, respectively, in the imidazole ring. On the other hand, 5-substituted compound 4f showed a pair of singlets at δ 7.10 and 7.92 ppm, corresponding to H-4 and H-2, respectively. Shirley and Alley¹¹ reported that lithiation of 1-substituted imidazole with n-butyllithium resulted in lithiation at the 2 position, and did not give a 5-substituted compound. The direct lithiation¹² of benzothiazole or 1-benzylbenzimidazole gave only the bis compounds 6b and 6c, respectively.

A similar reaction of 2,3-O-isopropylidene-D-ribono-1,4lactone (7) (Chart II) with benzothiazole of 1-benzylbenzimidazole gave 1-(benzothiazol-2-yl)-2,3-O-isopropylidene- β -D-ribofuranose (8b) and 1-(1-benzylbenzimidazol-2-yl)-2,3-O-isopropylidene- β -D-ribofuranose (8c), and 8b was acetylated to the 1,5-di-O-acetyl derivative (9b) in a usual manner. The reaction of sugar lactones with lithiated het-



erocycles progressed stereospecifically and isomeric lactols were not detected by thin layer and gas chromatography. Reductive elimination of 1-benzylbenzimidazol-2-yl derivatives (4c, 8c) over palladium on charcoal in a hydrogen atmosphere afforded the corresponding benzimidazolyl derivatives (4d, 8d) in a good yield (70 and 65%, respectively).







Figure 4. CD (---) and ORD (----) curves in methanol at 28°.

Chilton and Krahn,¹³ and Moffatt, et al.,¹⁴ already reported the relationship between the absolute configuration at the C-1 position of sugar heterocycles, such as benzimidazole, quinoxaline, flavazole, and anhydroosazone derivatives, and the Cotton effect of the ORD. Satoh, et al.,15 also reported that the absolute configuration of 1-nitroheptitols was determined by ORD and CD. Snatzke and his coworkers¹⁶ had reported the CD spectra of benzothiazole and benzothiazoline derivatives on aldoses and its acetates. Previously, we also reported¹² the Cotton effect and the structural relation of 1,2-dideoxy-4,5:7,8-di-O-isopropylidene-1-phenyl-L-glycero-D-galacto-oct-1-ynitol and 1,2-dideoxy-4,5-O-isopropylidene-1-phenyl-p-allohept-1ynitol. Moreover, the ethynylation reaction of sugar lactones 3 and 7 with lithiated ethynyl compounds using nbutyllithium or by the direct lithiation resulted in the attack of the reagent from the less hindered face to form the β -C-nucleoside.^{6,12}

The stereochemistry of the gulofuranosyl derivatives 4a-g and ribofuranosyl derivatives 8b-d was confirmed by the Cotton effect of CD curves of ring-opened alcohols, which were formed by sodium borohydride reduction.

As shown in Figures 1-3, the positive Cotton effect was observed from CD and ORD curves of gulofuranosyl com-





pounds (4a-c) and 2,3:5,6-di-*O*-isopropylidene-1-(2-substituted)hexane-L-glycero-D-galacto-1,2,3,4,5,6-hexol (11a,b,c,g) (Chart III). From this result, gulofuranosyl



compounds and L-glycerohexol compounds should have the β configuration at the 1 position; therefore compounds **4a,c,g** and **11a,c,g** should have S chirality at the 1 position and R chirality for **4b** and **11b**.

On the other hand, the negative Cotton effect was observed in the ribonosyl compounds 8b,c and 2,3-O-isopropylidene-1-(2-substituted)pentane-D-altro-1,2,3,4,5-pentol (12b,c) (Figures 4-6). From these results, ribonosyl compounds and D-altropentol should have the β configuration

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Figure 5. CD curves in methanol at 30°.





Figure 6. CD curves in methanol at 30°.

at the 1 position; therefore compounds 8c and 12c should have the R chirality at the 1 position and S chirality for compounds 8b and 12b.

In conclusion, asymmetric synthesis of C-nucleoside analogs was effected via lithiated heterocycles by a onestep synthesis. The absolute configuration of the products was confirmed from the Cotton effect. Ethynylation of isopropylidenegulonolactone (3) or -ribonolactone (7) afforded lactols which have the same chirality at the anomeric position.^{6,12} This conclusion differs from the results of Cnucleoside analogs, in which gulofuranosyl derivatives (4) obtained had the same stereochemistry with ethynyl derivatives: S chirality (4a,c,g) and R chirality (4b) at the anomeric position. However, ribofuranosyl derivatives (8) have 'R chirality (8c,d) and S chirality (8b) at the anomeric position.

Attempted elimination of the anomeric hydroxyl group failed. When 1-benzylbenzimidazolylgulonolactol (4c) was treated with formic acid in trimethylamine or sodium carbonate in formic acid, only 1-benzylbenzimidazole was obtained. Treatment of 4c with thionyl chloride-pyridine or phosphoryl chloride-pyridine also resulted in cleavage of the carbon-carbon bond to form 1-benzylbenzimidazole and sugar lactone (3). Hydrogenolysis of 4c with lithium aluminum hydride-aluminum chloride at room temperature gave many products on tlc, and the desired product was not obtained.

Experimental Section¹⁷

Reaction of 1-Phenylethynyl-2,3-O-isopropylidene- α -D-ribofuranose (1) with Benzylsydnone. A solution of 1 (160 mg, 0.55 mmol) and benzylsydnone or N-nitrosobenzylglycine (106 mg, 0.55 mmol) in acetic anhydride (10 ml) was heated under reflux for 8 hr. When cooled, the reaction mixture was poured into icewater and extracted with chloroform. Evaporation of dried chloroform solution left a brown syrup, which was chromatographed over silica gel and eluted with hexane-benzene. There was obtained 70 mg (26%) of an unidentified compound (2) as colorless needles: mp 125-127°; ir (KBr) 2180 (C=C), 1380, 1370 (CH₃), 1590, 765 cm⁻¹ (phenyl); nmr (CDCl₃) δ 4.47 (2 H, dd, =CH₂), 4.78 (1 H, d, H-1), 1.38, 1.58 ppm (6 H, s, isopropylidene).

Anal. Calcd for $C_{16}H_{16}O_3$: C, 74.98; H, 6.29; m/e 256.110. Found: C, 74.95; H, 6.30; m/e 256.108 (M⁺).

General Procedure for Preparation of 2,3:5,6-Di-O-isopropylidene- β -L-gulofuranosyl Derivatives (4a-c, e-g) (Table I) and 2,3-O-Isopropylidene- β -D-ribofuranosyl Derivatives (8b,c)

Table I	
2,3 : 5,6-Di-O-isopropylidene- β -L-gulofuranosyl Derivatives β	(4a–g)

Compd	R	Yield, %	Mp, °C	Formula	c	alcd, % H	{ N	C F	ound, %— H	N	-Mass, m Calcd	ı/e (M⁺)— Found
4a	Pyrid-2-yl	74ª	86-87	$C_{17}H_{23}NO_{6}$	60.52	6.87	4.15	60.69	6.48 3.	89	322.129	322.129°
4b	Benzothiazol-2-yl	56	170 - 171	$C_{19}H_{23}NO_6S$	58.01	5.89	3.56	58.10	5.96 3.	47	393.123	393.125
4c	1-Benzylbenzimidazol-2-yl	40	155 - 156	$C_{26}H_{30}N_2O_6$	66.93	6.48	6.01	66.68	6.68 6.	10	466.210	466.210
4 d	Benzimidazol-2-yl	70	175 - 176	$C_{19}H_{24}N_2O_6$	60.63	6.43	7.44	60.94	6.50 7.	47	376.163	376.163
4e	1-Benzylimidazol-2-yl	30	87-88	$C_{22}H_{28}N_2O_6$							416.195	416.193
4f	1-Benzylimidazol-5-yl	12	199 dec	$C_{22}H_{28}N_2O_6$	63.44	6.78	6.73	63.15	6.95 6.	69	416, 195	416, 196
4g	Benzylsydnon-4-yl	29 32 ⁶	157–158	$C_{21}H_{26}N_2O_8$	58.06	6.03	6.45	57.71	6.06 6.	23	434.169	434.165

^a From α -bromopyridine. ^b From 3-benzyl-4-bromosyndnone. ^c (M - CH₃) +.

Table II			
2.3-O-Isopropylidene-B-D-ribofuranosyl Derivative	s (8b-d	and	9b)

Compd	R	Yield, %	Mp, °C	Formula	Calcd	$n/e (M^+) - Found$
8b	Benzothiazol-2-yl	23	110-111	C ₁₅ H ₁₇ NO ₅ S	323.083	323.083
8c	1-Benzylbenzimidazol-2-yl	25	175 - 176	$C_{22}H_{25}N_{2}O_{5}$	396.169	396,169
8d	Benzimidazol-2-yl	65	173 - 174	$C_{15}H_{18}N_{2}O_{5}$	306.122	306.123
9b	1,5-Di- <i>O</i> -acetyl 8a	32	166 - 167	$C_{19}H_{21}NO_7S$	407.104	407.104

Table III1-O-Acetyl-2,3: 5,6-Di-O-isopropylidene- β -L-gulofuranosyl Derivatives (5a-c)

Compd	R	Yield, $\%$	Mp, °C	Formula	Calcd	m/e (M+) Found
5a	Pyrid-2-yl	24	147–148 dec	$\begin{array}{c} C_{19}H_{25}NO_{7}\\ C_{21}H_{25}NO_{7}S\\ C_{28}H_{32}N_{2}O_{7} \end{array}$	379.163	379.165
5b	Benzothiazol-2-yl	36	163–164		435.135	435.129
5c	1-Benzylbenzimidazol-2-yl	24	176–177		493.197	493.198°

^a $(M - CH_3)^+$.

Table IV β -L-Gulitols (11a-c, g) and β -D-Ribitols (12b,c)

Compd	R	Yield, $\%$	Mp, °C	Formula	C C	alcd, % H	N	C F	ound, 9 H	% N	—Mass, m Calcd	$e (M^+)$ Found
11a 11b 11c 11g	Pyrid-2-yl Benzothiazol-2-yl 1-Benzylbenzimidazol-2-y Benzylsydnon-4-yl	54 57 1 40 42	84–85 97–98 178–179 148–149	$\begin{array}{c} C_{17}H_{25}NO_6\\ C_{19}H_{25}NO_6S\\ C_{23}H_{32}N_2O_6\\ C_{21}H_{28}N_2O_8 \end{array}$	60.16 57.71 66.65 57.79	4.13 6.37 6.88 6.47	7.43 3.54 5.98 6.42	59.82 57.65 66.43 57.81	3:94 6.65 7.09 6.62	7.59 3.25 5.72 6.38	339.168 395.140 468.222 421.161	339.168 395.140 468.225 421.158
12b 12c	Benzothiazol-2-yl 1-Benzylbenzimidazol-2-y	70 1 54	$64-65 \\ 76-78$	${{ m C}_{15}}{{ m H}_{19}}{ m NO}_5{ m S} \ {{ m C}_{22}}{{ m H}_{26}}{ m N}_2{ m O}_5$	55,37 66,32	5.89 6.58	4.30 7.03	55,35 66,33	5.87 6.59	$\begin{array}{c} 4.32 \\ 7.01 \end{array}$	325.098 398.183	325.098 398.184

Table VAcetates of 11b and 12b

Compd	Yield, %	Mp, °C	Formula	C	alcd, % H	N	C F	ound, % H	N	Calcd	Mass, m/e
Acetyl β -L-gulitol Acetyl β -D-ribitol	15 30	95–96 129–130	$C_{23}H_{29}NO_8S$ $C_{21}H_{25}NO_8S$	57.61 55.87	6.10 5.58	2.92 3.10	57.72 55.90	6.33 5	2.66	479.161 436_107	$479.161 (M^+)$ $436.106 (M^- CH_3)^+$

(Table II). To an ether solution of *n*-butyllithium prepared from lithium (0.2 g, 0.03 mol) and *n*-butyl bromide (2.5 g, 0.02 mol), the heterocyclic compound (0.01 mol) in ether (5-10 ml) was added slowly during 20-30 min at below -70° . After the reaction solution was stirred for 2 hr at room temperature, 2,3:5,6-di-O-isopropylidene- γ -L-gulonolactone (3, 2.5 g, 0.01 mol) in freshly distilled tetrahydrofuran (10-20 ml) was added dropwise into the cooled reaction solution, and the stirring was continued for 2-3 hr. The reaction mixture was then allowed to stand overnight at room temperature. The reaction mixture was treated with saturated ammonium chloride solution and extracted with ether (150 ml). The organic layer was washed with water and dried over magnesium sulfate. The extracts were concentrated under reduced pressure to give 4a-c,e-g.

Acetylation of 4a-c (Table III). The compounds 4a-c were acetylated with acetic anhydride (4 ml) and pyridine (5 ml). The reaction solution was stirred for 20-30 hr at room temperature and poured into ice-water. The solution was extracted with chloroform and the organic layer was washed with saturated sodium bicarbonate solution and water. The extract was dried and concentrated under reduced pressure.

Reductive Elimination of Benzyl Group from 4c and 8c (Ta-

bles I and II). A methanol (60 ml) solution of 4c or 8c (0.01 mol) was hydrogenated over 5% palladium on charcoal. After the reaction; the filtered solution was concentrated under reduced pressure to give 1-(2-benzimidazolyl)-2,3:5,6-di-O-isopropylidene- β -L-gulofuranose (4d) or 1-(2-benzimidazolyl)-2,3-O-isopropylidene- β -D-ribofuranose (8d).

Reduction of 4a-c,g with Sodium Borohydride (Table IV). The compound (4a-c,g, 0.01 mol) was dissolved in methanol (10 ml), and sodium borohydride (0.15 g) was added. After the reaction solution was stirred for 2-24 hr at room temperature, excess reagent was decomposed with ethyl acetate and water, and the organic layer was washed with 0.1 N hydrochloric acid and water, dried, and evaporated to give 11a-c,g.

Reduction of 8b,c with Sodium Borohydride (Table IV). After a similar procedure as above, the obtained syrup was chromatographed over silica gel with hexane-chloroform (85:15) and the desired product was isolated as an oily product.

Acetylation of 11b and 12b (Table V). The compound (11b, 12b) was acetylated with acetic anhydride in pyridine, and the acetate was obtained after chromatography and recrystallization from chloroform-ether.

Attempted Elimination of Tertiary Hydroxyl Group. A. With

Formic Acid. A solution of 4c (50 mg) in 5 ml of trimethylammonium formate [bp 92° (18 mm)] was stirred for 2 hr at room temperature and left overnight. This solution was gently refluxed in an oil bath for 3 hr until the reaction mixture was colored dark brown. When cooled, the separated crystals were collected and recrystallized from hexane-chloroform to colorless needles, mp 195-196°. This was treated with 1 N sodium hydroxide form 1-benzylbenzimidazole, mp and mmp with authentic sample 115°

B. With Phosphoryl Chloride or Thionyl Chloride in Pyridine. To a solution of 4c (400 mg) in pyridine or pyridine-chloroform, phosphoryl chloride (5 ml) or thionyl chloride (4 ml) was added at 0-5°. After the reaction mixture was stirred for 1 hr at room temperature, it was poured into ice-water. Extraction of the reaction mixture with benzene afforded 1-benzylbenzimidazole.

C. With Lithium Aluminum Hydride-Aluminum Chloride. To an ether solution of 4c (450 mg), lithium aluminum hydride (50 mg) and aluminum chloride (25 mg) were added under stirring at 0-5°. After stirring overnight at room temperature, the reaction mixture was treated with ethyl acetate and then with 0.1 N hydrochloric acid. Evaporation of the dried ether solution left a brownish syrup, which showed four spots on tlc ($R_{\rm f}$ 0.79, 0.45, 0.30, and 0.14), and the main spot $(R_f 0.45)$ was found to be 1benzylbenzimidazole.

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Synthesis of Macrolide Antibiotics. I.¹ Stereospecific Addition of Methyllithium and Methylmagnesium Iodide to Methyl α -D-xylo-Hexopyranosid-4-ulose Derivatives. Determination of the Configuration at the Branching Carbon Atom by Carbon-13 Nuclear Magnetic Resonance Spectroscopy

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Methyllithium (LiBr-free) adds stereospecifically to methyl 2,3-di-O-methyl-6-O-triphenylmethyl- α -D-xylohexopyranosid-4-ulose (1) and methyl 3-O-methyl-2-O-methylsulfonyl-6-O-triphenylmethyl- α -D-xylo-hexopyranosid-4-ulose (2) in an ethereal solution at -80° to give methyl 2,3-di-O-methyl-4-C-methyl-6-O-triphenylmethyl- α -D-glucopyranoside (9) and methyl 3-O-methyl-4-C-methyl-2-O-methylsulfonyl-6-O-triphenylmethyl- α -D-glucopyranoside (11), respectively. Methylmagnesium iodide adds to the oxo sugars 1 and 2 in an ethereal solution at -80° again stereospecifically, giving methyl 2,3-di-O-methyl-4-C-methyl-6-O-triphenylmethyl- α -Dgalactopyranoside (8) and methyl 3-O-methyl-4-C-methyl-2-O-methylsulfonyl-6-O-triphenylmethyl-α-D-galactopyranoside (10), which are, however, the C-4 epimers of the branched-chain sugars 9 and 11. The stereochemistry of the addition of Grignard reagent to the oxo sugars 1 and 2 depended upon the reaction temperature, the solvent, and the nature of the halogen atom. Carbon-13 nmr spectroscopy was used for unequivocal configurational assignments at the branching-carbon atom in branched-chain sugars 8-11. A rationalization of the observed stereospecificity was proposed.

In the course of our studies directed toward the stereoselective synthesis of the 14-membered lactone ring of ervthromycins A and B from appropriate sugar derivatives, it was necessary to introduce an axial methyl group at the C-4 carbon atom of a methyl p-xylo-hexopyranosid-4-ulose derivative and to develop a simple but reliable method for configurational assignment of the thus obtained branching carbon atom.²

It is well known that the addition of Grignard reagents and organolithium compounds to carbonyl groups in carbohydrates is highly stereoselective⁴ yielding in certain cases products epimeric at the quaternary carbon atom,^{5,6} whereas in other instances branched-chain sugars with the same configuration at the branching carbon atom⁷ are obtained. Since a clear rationalization of these findings⁸ does not exist, many stereochemical "anomalies"⁴ re-